AS 1329.7—1994 Reconfirmed 2016

## Australian Standard®

Methods for the analysis of zinc and zinc alloys

Part 7: Determination of lead content—Flame atomic absorption spectrometric method This Australian Standard was prepared by Committee CH/10, Analysis of Metals. It was approved on behalf of the Council of Standards Australia on 28 June 1994 and published on 19 September 1994.

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Aluminium Development Council Australasian Institute of Mining and Metallurgy Australian Lead Development Association Bureau of Steel Manufacturers Copper Technical Data Centre National Association of Testing Authorities, Australia Railways of Australia Committee

Additional interests participating in preparation of Standard:

Analytical laboratories Department of Defence, Materials Research Laboratory Steel manufacturers

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#### RECONFIRMATION

#### OF

## AS 1329.7—1994 Methods for the analysis of zinc and zinc alloys Part 7: Determination of lead content—Flame atomic absorption spectrometric method

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Approved for reconfirmation in accordance with Standards Australia procedures for reconfirmation on 31 July 2016.

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Australian Aluminium Council Bureau of Steel Manufacturers of Australia International Copper Association Australia International Precious Metals Institute National Association of Testing Authorities Australia NOTES

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Part 7: Determination of lead content—Flame atomic absorption spectrometric method

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## PREFACE

This Standard was prepared by the Standards Australia Committee CH/10 on the Analysis of Metals to supersede AS 1329.7—1980, *Methods for the analysis of zinc and zinc alloys*, Part 7: *Determination of lead content*—*Flame atomic absorption spectrometric method*.

## CONTENTS

Paga

		uge
1	SCOPE	. 3
2	REFERENCED DOCUMENTS	. 3
3	PRINCIPLE	. 3
4	REAGENTS	. 3
5	APPARATUS	. 4
6	SAMPLING	. 4
7	PROCEDURE	. 4
8	DETERMINATION OF RESULTS	. 6
9	PRECISION	. 6
10	ACCEPTANCE OF ANALYTICAL VALUES	. 7
11	TEST REPORT	. 8

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## STANDARDS AUSTRALIA

## **Australian Standard**

## Methods for the analysis of zinc and zinc alloys

# Part 7: Determination of lead content—Flame atomic absorption spectrometric method

**1 SCOPE** This Standard sets out a flame atomic absorption spectrometric method for the determination of the lead content of special high grade zinc and high grade zinc, and in diecast alloys containing a maximum of 4.3% aluminium, 0.06% magnesium and 1.3% copper. The method is suitable for the determination of lead content in the range 0.0001% to 0.03%.

**2 REFERENCED DOCUMENTS** The following documents are referred to in this Standard:

AS

2134 Recommended practice for analysis by atomic absorption spectrometry

- 2134.1 Part 1: Flame atomic absorption spectrometry
- 2162 Code of practice for the use of volumetric glassware
- 2164 One-mark volumetric flasks
- 2165 Burettes and bulb burettes
- 2166 One-mark pipettes
- 2347 Method for the sampling of zinc metal and zinc alloys for chemical analysis
- 2850 Chemical analysis—Interlaboratory test programs—For determining precision of analytical method(s)—Guide to the planning and conduct
- BS
- 4237 Report on reproducibility of methods of chemical analysis used in the iron and steel industry

**3 PRINCIPLE** The sample is dissolved in sulfuric acid/nitric acid mixture. Sodium iodide is added and the lead iodide complex is extracted into n-butyl acetate containing 1% of tri-noctylamine. The lead is determined in the organic phase by flame atomic absorption spectrometry.

## **4 REAGENTS**

**4.1 General requirements** During the analysis, only reagents of recognized analytical reagent grade, and only distilled water or water of equivalent purity, shall be used.

## 4.2 Solutions

**4.2.1** Sulfuric acid/nitric acid mixture Add, slowly and with stirring, 150 mL of sulfuric acid ( $\rho_{20}$  1.84 g/mL) and 30 mL of nitric acid ( $\rho_{20}$  1.41 g/mL) to 700 mL of water. Cool, dilute to 1 L, and mix.

**4.2.2** Sodium iodide solution (750 g/L) Dissolve 75 g of sodium iodide and 4 g of ascorbic acid in water. Dilute to 100 mL and mix.

NOTE: Potassium iodide solution (830 g/L) may be used in lieu of sodium iodide solution.